

**A LETTER TO MR.
EULER, PROFESSOR
OF PHILOSOPHY,
AND MEMBER OF
THE IMPERIAL...**

Benjamin Wilson



different from your's, which promises to confirm the philosophy of Sir Isaac Newton. But, before I enter upon those matters, it may be proper, on many accounts, to state a few general facts, together with such singular phosphoric appearances, as seem the most difficult to be reconciled with true philosophy.

By a great variety of experiments upon Phosphori I found, that I could, by the particular mode of preparation, make a phosphorus exhibit any particular colour.

But while I was engaged in those interesting experiments, I recollected to have seen a paper of Father Beccaria's (of Turin) upon the subject of Phosphori, in the Philosophical Transactions for the year 1771, which tended to prove, by means of coloured glasses, that *Phosphori emitted the same light as they received*.

This fact appearing so very extraordinary, compared with what I had at that time experienced in my phosphoric researches, I set about trying the effect of the primitive rays themselves upon different coloured Phosphori, after they were refracted by a prism, and properly intercepted, so that only a single ray should enlighten the phosphorus. When to my great surprize I found, that the phosphorus which had been prepared to give a *red light* did, after being exposed to a *red ray*, exhibit only a very *faint red light*; and that the same phosphorus did, after being exposed to a *violet ray*, exhibit a very *bright red light*; that a *blue ray* made it shine, but not quite so brilliant; and a *green ray* still less.

I also found, that a *blue phosphorus* did, after exposing it to a *blue ray*, exhibit in the dark a very *faint blue light*. But this was not the case when the same blue phosphorus had been exposed to a *red ray*; for then it shone with a *fine blue light*.

In consequence of these and other experiments I find Father Beccaria, in a printed letter addressed to me, hath expressed himself, that he must have been deceived in his experiments; because (says he) "*I have taken for very light tints of colours (which only could be expected) what was merely a faint gradation of light, or a different degree of shade.*" *

Thus

* "Io debbo convenire d'essermi ingannato nella speranza mia forse perchè ho preso per tinta di colori leggerissimi (quale unicamente si doveva aspettare) ciò che era unicamente una varia degradazione di luce, o una diversa maniera d'ombra."

cessary, that you might be acquainted from whence the discovery, which I have now been relating, took its rise.

The cause of those curious appearances I could not, at that time, even guess at, but conceived they might depend upon certain properties of light which had hitherto escaped the observation of Philosophers: what my opinion is now upon this subject, will presently appear.

It is three years since the second edition of my book was published, in which those remarkable experiments were first mentioned: during that period I never met with any one who had attempted to explain the cause of them, 'till I had the honour of receiving your memoir. And although I read it with the greatest admiration, yet I must confess, that there are some things in which I cannot agree in opinion with the truly learned and excellent author.

By the following passage, in your account of my experiments, viz. * “ So that any phosphoric red body having been enlightened by red rays, and afterwards *carried* into the dark,” I find you have misunderstood me in a material circumstance. You have conceived that the phosphorus was carried into another room, to be observed, after it had been enlightened: but that was not the fact; the experiments were made in the same dark room, into which a beam of the sun's light was occasionally introduced and separated by a prism. Whenever therefore I wanted to enlighten a phosphorus with one ray of light, I intercepted all the others. Immediately after which, an assistant shut out the beam of light, and at that instant, or nearly so, I viewed the phosphorus; but my eyes for some time before that instant, were always shut, merely for the purpose of seeing better when the experiment required them to be open.

I have been the more particular in this account, because you seem to lay great stress upon some difference in point of time. For in another part of the same memoir you say, “ We need only “ to distinguish carefully two moments in Mr. Wilson's experiments.

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* Ainsi un tel corps phosphorique rouge ayant été éclairé par les rayons rouges, et ensuite transporté dans l'obscurité.

ments. * The first of which is the moment he enlightens his red phosphoric body by the violet rays of the sun, at which time there is no doubt but that the body must appear almost entirely destitute of all colour. It is only at the second moment when he carries this dark body into the chamber, that it begins to shine with its proper colour." You doubtless express yourself here by the words, "begins to shine," as if the phosphorus was incapable of shining at the first moment; that is, when I exposed it to the light. In reply to this very material part, I shall quote two or three passages from my phosphoric work, which I think are very much to the present purpose.

† "With regard to the shells exhibiting prismatic colours in the dark, and yet affording no other appearance in the open day, than the same white light they acquired by calcination, a very little consideration may, perhaps, explain this matter, as we find it so exactly consonant with other phenomena in the natural world. The stars, and even the moon itself, are totally obscured when the sun is in the meridian, by the flood of light that is diffused over the whole of our atmosphere. And yet, when that superior light is withdrawn, how very splendid do the moon and stars appear! Now the darkness we produce by art, exceeds this natural one by many degrees. No wonder then, that the difference between the external light, and the closet we have described, is far greater than that between noon and midnight; and more especially whilst the moon continues shining. It is, therefore, entirely owing to this very great difference, that the prismatic colours in the shells become so visible in the dark, when the eye is properly prepared. So likewise, those colours, though they actually continue to exist

† Page 67 and 68.

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* "On n'a qu'à bien distinguer deux momens dans les expériences de M. de Wilson, dont le premier est celui où il a éclairé son corps phlogistique rouge par les rayons violets du soleil, pendant lequel il n'y a aucun doute, que le corps n'ait paru presque entièrement dénué de toute couleur : ce n'étoit que dans le second moment lorsqu'il transportoit ce corps dans l'obscurité, qu'il a commencé à briller de sa propre couleur rouge."

“ to the sun, by the very great quantity of reflected light with
“ which they are surrounded.”

“ And however strange it may appear, I have some foundation for
“ apprehending, that the splendor of the prismatic colours in the
“ open day, could we perceive them, is far greater than that we
“ observe in the dark. This notion arises from many observations
“ that were made during the whole course of the preceding experiments.
“ For when the different phosphori were brought from
“ the light into the dark, the splendor of the colours was exceedingly
“ more vivid in the first instant, than it appeared in the third
“ of a second afterwards. The disappearing of this great degree of
“ brilliancy was so sudden, that I could seldom move the substances,
“ which exhibited it, a few inches from the curtains, before a considerable
“ alteration in its vividness had taken place; notwithstanding the time in moving the phosphori from the curtains to the
“ convenient distance for observing, (it being about 12 or 15 inches)
“ was not more than one second.”

In another part, (page 115) where I excited phosphoric appearances by means of the electric light, which was caused by the discharge of a large Leyden Jar, are these words:

“ — Besides, by this method of exciting prismatic colours, I obtained another advantage, which appears to me not inconsiderable.
“ For here, the colours may be seen at the instant, after the explosion hath taken place; and, agreeably to a former observation,
“ (meaning the above) they appear much more full, intense, and
“ lively than in three or four seconds afterwards; or even after they
“ have been exposed to the light of the sun.”

When I published the preceding observations I had framed no hypothesis: consequently there was not any inducement at that time which could influence my judgment.

Now as those observations certainly tend to prove that phosphori actually shine at the time they are exposed to light, but are rendered invisible by the presence of a superior light, and are only visible when that superior light is withdrawn; do not those observations materially interfere with your manner of reasoning,
which

which is entirely founded upon the different effects taking place at the two different moments you have marked?

That this is the true state of the case will appear more fully by the following quotation taken from the *two last pages* of your memoir.

* “ But to come to what relates more particularly to these new experiments, I have demonstrated that it is absolutely false, “ that

* “ Mais pour ce que regarde plus près les nouvelles expériences, j'ai démontré, “ qu'il est absolument faux, que les corps opaques nous deviennent visibles par des “ rayons réfléchis comme on s'étoit généralement imaginé autrefois, mais qu'il faut “ absolument, que les moindres particules qui se trouvent dans la surface de ce corps “ soient mises dans un certain mouvement de vibration plus ou moins rapide selon “ que la couleur du corps l'exige, attendu qu'à chaque couleur il répond un certain “ nombre de vibrations achevées pendant une seconde. Car alors ce même mouvement “ produit des semblables vibrations dans l'éther environnant, d'où résultent des rayons “ de la même couleur. Par là il est clair que les rayons, par lesquels nous voyons “ les corps opaques, sont engendrés dans leur propre surface conformément au degré “ d'élasticité dont les moindres particules y sont douées. Or pour mettre ces parti- “ cules dans un tel mouvement de vibration, il faut que des rayons de lumière y “ tombent, qui par leur action les excitent à un tel mouvement; de la même “ manière qu'un corde de musique en repos étant exposée à un son assez fort com- “ mence à trembler et à rendre elle-même un son qui répond à son degré de tension.

“ Cela posé un corps phosphorique rouge tel que Mons. Wilson a examiné, ne “ nous sauroit devenir visible, qu'en tant que les moindres particules dans la surface “ ne soient excitées à un mouvement de vibration qui convient à sa propre couleur. “ Ce seront donc sans doute les rayons rouges, qui seront les plus propres à imprimer à “ ces particules un tel mouvement de vibration, qui par la nature phosphorique de “ ce corps se conserveront encore pendant quelque tems après que les rayons incidents “ auront cessé d'y agir, mais d'une manière beaucoup plus foible que pendant que “ les rayons rouges y ont agis actuellement. Voyons à présent quel effet les rayons “ violets doivent produire sur ce même corps phosphorique rouge, et d'abord il est “ évident qu'ils ne sauroient porter ses moindres particules à un mouvement de “ vibration à cause de la contrariété qui regne entre les vibrations des rayons violets “ et celles que les propres particules du corps sont disposés à recevoir. Par cette “ raison tout l'effet de ces rayons violets se réduira à pousser les particules du corps “ à un certain degré de tension sans leur imprimer un mouvement actuel. Donc “ aussitôt que le corps sera retiré de l'action de ces rayons, ses moindres particules “ commenceront à se dégager de leur état de tension, et recevront le même mouve- “ ment de vibration qui est propre à leur nature, et par tant elle répandront des “ rayons rouges, qui seront même plus forts, à cause du haut degré de tension, que “ si le même corps avoit été exposé aux rayons rouges. Enfin par la nature des “ corps phosphoriques ce mouvement de vibration pourra durer plus ou moins long “ tems selon le degré dont ces corps seront doués de la qualité phosphorique.”

“ be put into a certain movement of vibration, more or less
“ rapid, according as the colour of the body requires: since to
“ each colour there corresponds a certain number of vibrations to
“ be performed during a second: for then this same movement
“ produces similar vibrations in the surrounding æther. From
“ thence it is clear, that the rays, by which we see opaque bodies,
“ are *engendered* in their proper surfaces, conformably to the degree
“ of elasticity with which their smaller particles are endued.
“ Now to put these particles into such a movement, rays of light
“ must fall on them, which by their action may excite them to
“ such a movement. In the same manner as a musical chord at
“ rest being exposed to a sound sufficiently strong, begins to
“ vibrate; and to give a sound which corresponds to its degree of
“ tension. This being laid down, a red phosphorical body, such
“ as Mr. Wilson has examined, can only become visible to us in
“ consequence of the smaller particles in its surface, being excited
“ to a movement of vibration which answers to its proper colour:
“ therefore, doubtless, the red rays will be the most proper to com-
“ municate to those particles such a movement of vibration, as
“ from the phosphorical nature of this body, will still continue
“ for a certain time, after that the incident rays shall have ceased
“ to act upon it; but in a manner much more feeble, than whilst
“ the red rays were immediately acting upon it.

“ Let us now consider what effect the violet rays ought to pro-
“ duce upon this same phosphorical body. In the first place it is
“ evident that they cannot dispose its smaller particles to a move-
“ ment of vibration, upon account of the contrariety which reigns
“ between the vibrations of the violet rays, and those which the
“ proper particles of the bodies are adapted to receive. From this
“ reason the whole effect of these violet rays will be reduced to
“ forcing the particles of the body to a certain degree of tension;
“ without communicating to them any actual movement (of vi-
“ bration). But as soon as the body shall be withdrawn from the
“ action

“ action of those rays, its smaller particles *will begin to disengage themselves from their state of tension*; and will receive that movement of vibration which is proper to their nature: and in consequence will emit red rays: which will be even stronger, upon account of the high degree of tension, than if the same body had been exposed to red rays. Lastly, from the nature of phosphorical bodies, this movement of vibration will be enabled to continue during a longer or a shorter time, according to the degree of phosphorical quality, with which the bodies are endued.”

Thus far Mr. Euler.

LET us now see how far those singular phosphoric phænomena may be explained upon other principles:

When a beam of the sun's light is let fall upon any coloured phosphorus, the ray, which is of the same colour with the phosphorus, will be always most reflected: while the others are absorbed, in proportion to their difference in the order of colours from that reflected ray.

For instance, if the beam fall upon a red phosphorus, the red ray will be most reflected, and the other rays absorbed, in proportion to their difference from the red one.

Again, if it be let fall upon a violet phosphorus, the violet ray will be most reflected, and the other rays absorbed in like manner.

Now these rays that are so absorbed will, by refractions and reflections within the phosphoric body, excite those vibrations which cause it to emit its proper light.

For example, the *violet rays* being absorbed in the red phosphorus will, by refractions and reflections, excite vibrations, and cause it to emit a *brilliant red light*, which is the proper colour of the phosphorus.

In like manner the *blue rays*, which are less absorbed than the violet, will excite a red light not quite so brilliant, and the green rays less brilliant still. So, for the same reason, the *red rays*, being absorbed in the violet phosphorus, will excite vibrations, and cause it to emit a *brilliant violet light*, which is also the proper colour of the phosphorus.

Lastly,

and cause it to emit a violet light, but not quite so brilliant. And the green rays, being still less absorbed, will excite weaker vibrations, and cause the violet phosphorus to emit a fainter light.

All which effects seem to be perfectly consistent with Sir Isaac Newton's philosophy of light and colours *, and admit of an easy and natural solution, without having recourse to any other supposed properties of light, not known to that truly great Philosopher.

I am, with the greatest respect,

Great Russell-Street, Bloomsbury,
London, May 25, 1779.

Learned S I R,
Your most obliged humble servant,

B. W I L S O N.

Note, The learned Mr. Beguelin, I find, has presented to the Academy at *Berlin*, his opinion on my work upon Phosphori. It is inserted in the 7th vol. of the new *Berlin* Memoirs, where he expresses himself in the following words :

† “ Mr. Wilson, from an apprehension of shaking the doctrine
“ of the immutability of the coloured rays, explains the obser-
“ vation of the 7th experiment by the supposition, that the solar
“ rays, of what kind soever, have the power of kindling a light
“ flame in the parts of Phosphori. And that this flame has the
“ colour of the phosphorus itself, and not the colour of the kind-
“ ling ray. But this explication does not afford a reason why the
“ weaker rays, such as the blue and the violet, excite a stronger
“ flame than the red or the yellow rays.”

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* Part 2d, Book 1, prop. 10.
Part 3d, Book 2, prop. 9. and 10.

† “ Monf. Wilson, dans la crainte de porter atteinte à la doctrine de l'immu-
“ bilité des rayons colorés, explique l'observation de l'expérience VII. par la sup-
“ position que les rayons solaires quelconques ont le pouvoir d'allumer une legere
“ flamme dans les parties des phosphores, et que cette flamme a la couleur propre
“ au phosphore même, et non la couleur du rayon allumant ; mais cette explication
“ ne rend pas raison pourquoi les rayons les plus foibles, tels que les bleux et les
“ violets, excitent une plus forte flamme que n'excitent les rayons rouges, ou les
“ jaunes.”

I think that this difficulty of Mr. Beguelin's, which might seem to contradict, or at least to weaken the system of Sir Isaac Newton, is already compleatly removed in the above memoir.

To the proofs which I have already given in support of the opinion that Phosphori shine by a flame kindled within their substance, may be added some very curious and striking experiments made at *London*, by the celebrated Monf. L'Abbé Fontana, Physician to the Grand Duke of *Tuscany*.

He has proved by direct experiments, that the Phosphorus of *Bologna*, that of Canton's, and that made with lime, being inclosed in a vessel full of common air, diminished that air sensibly, and rendered it phlogisticated. This he verified by nitrous air. Now it is certain, as he reasons, in consequence of these experiments, that those effects are always produced by substances in combustion, however insensible and small their flame may appear to be. The difference is only between the more or the less. The effects of the flame are the extrication of the phlogiston, the diminution and the phlogistication of the atmospherical air. It is not difficult to excite an actual flame in the greater part of bodies which actually produce phlogiston, only by augmenting a little the force of this principle, which causes the extrication of it.

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